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**PHYSICA B**

## SANS on fluorinated water-in-oil microemulsions

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### Abstract

Preliminary SANS results on water-in-oil microemulsions composed of water, fluorinated surfactant and fluorinated oil, indicate that the system is composed of interacting droplets in a large range of compositions. Work is in progress to find the interaction potential.

*Keywords:* Small-angle neutron scattering; Fluorinated microemulsion; Perfluoropolyether surfactant

Fluorinated water-in-oil microemulsions with perfluoropolyether compounds were recently characterised by phase diagram, conductivity, light scattering, nuclear magnetic resonance, dielectric spectroscopy, etc. (see Ref. [1] for a complete list of references). The ternary system water, fluorinated surfactant and fluorinated oil, show a large monophasic domain of homogeneous, transparent, isotropic samples [2]. Previous light scattering investigation [3] gave reliable results for the dilute microemulsions,  $\phi < 0.10$  where  $\phi$  is the volume fraction of the dispersed phase ( $\phi = (\text{water} + \text{surfactant})/\text{total}$ ) assuming the dispersed phase composed of water droplets coated by surfactant molecules. Water droplets on nanometer scale were identified at water to surfactant molar ratio ( $W/S$ ) higher than 6. The droplets maintain a constant radius at constant  $W/S$  ratio. Spherical shapes were hypothesized, as no experimental evidence of depolarized light was found and the polydispersity was very low (10%). The hydrodynamic radius increases versus the  $W/S$  ratio increase (27 Å at  $W/S = 6.5$ , 31 Å at 11, 44 Å at 16 and 57 Å at 22) for the most part of

hydrogenated water-in-oil microemulsions. The second virial coefficient  $\alpha$  ( $< -20$  at  $W/S = 6.5$ ,  $-8$  at 11,  $-2$  at 16 and  $\sim 0$  at 22) indicates that the attractive component of the interaction between droplets is higher for smaller droplets. This trend is opposite to that usually observed in hydrogenated water-in-oil microemulsions for which the attractive interaction between droplets is dominated by the attraction of water cores [4]. A percolation phenomenon mainly of dynamic nature was found in the system [1], representing further evidence that the microemulsion is composed of interacting droplets. The droplets give rise to a cluster of infinite size, or percolate, either for an increase of the number density of the aggregates themselves or for a temperature increase [1].

The fluorinated compounds [1] were from Ausimont S.p.A. (Milan, Italy). The water was from a Millipore Milli-Q system. SANS experiments were performed at the spectrometer PAXE (Lab. Léon Brillouin, Saclay) with a sample-detector distance of 2.5 m and incident neutron wavelength of 5 Å with wavelength spread of 10%. Collimation was achieved by two slits of 12 and 7 mm placed 2.5 m far apart. Samples of thickness 1 mm were contained in flat quartz cells, temperature controlled within  $\pm 0.1^\circ\text{C}$ .

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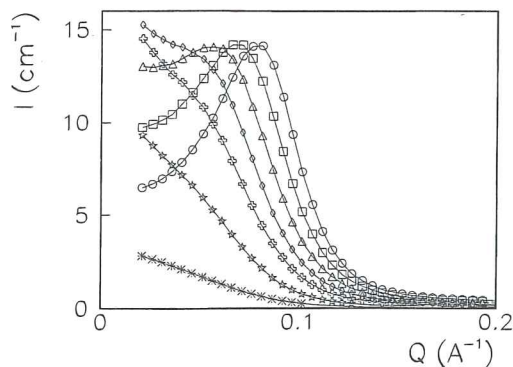


Fig. 1. Experimental scattered intensity at  $T = 20^\circ\text{C}$  of fluorinated microemulsions with  $W/S = 15$  and  $\phi$  values: 0.599 (circles), 0.499 (squares), 0.390 (triangles), 0.285 (rhombuses), 0.202 (crosses), 0.100 (stars) and 0.0303 (asterisks). The lines are guides for the eyes.

The intensity was corrected for the empty-cell contribution and normalized to absolute scale by means of a secondary standard of known cross-section [5].

In Fig. 1 the experimental scattered intensity as a function of  $Q$ ,  $I(Q)$ , is shown for samples with  $W/S = 15$  at  $T = 20^\circ\text{C}$  ( $\phi$  values in the range 0.60 to 0.03). For each curve, the contribution of the oil scattered intensity was subtracted. The principal characteristic of the spectra of Fig. 1 is the presence of a peak at  $Q \sim 0.08 \text{ \AA}^{-1}$  for  $\phi = 0.599$  and the shift of the peak to lower  $Q$  values as a function of the decrease of the volume fraction of the dispersed phase. The Guinier plot of the curve at  $\phi = 0.0303$  is reported in Fig. 2. A radius of the particle of  $35 \text{ \AA}$  is calculated, and found to be in good agreement with light scattering results. In neutron scattering, the fluorinated surfactant shell and the continuous fluorinated oily medium have the same scattering-length density thus it is reasonable that the Guinier radius is smaller than the hydrodynamic radius measured by light scattering. SANS investigation at  $W/S = 11$  gave a Guinier radius of  $23 \text{ \AA}$  and a similar interpretation was given in Ref. [6]. At  $W/S = 20$  a data set similar to that of Fig. 1 was detected; the Guinier radius of the dilute sample is  $36 \text{ \AA}$ , see Fig. 2. For comparison, the dilute sample at  $W/S = 5.6$  is reported in Fig. 2. The Guinier region is not observed.

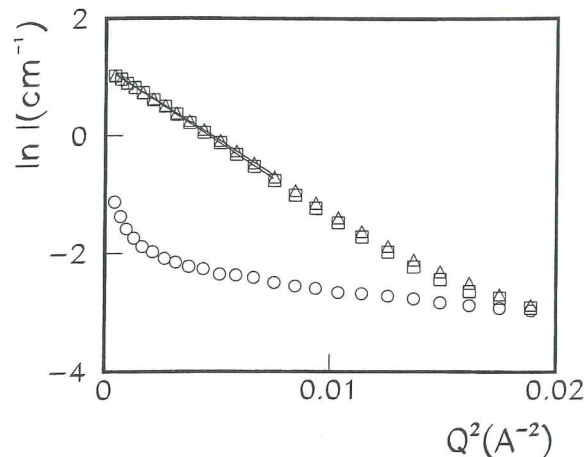


Fig. 2. Guinier plot of samples at  $W/S = 15$  with  $\phi = 0.0303$  (triangles), at  $W/S = 20$  with  $\phi = 0.0318$  (squares) and at  $W/S = 5.6$  with  $\phi = 0.0300$  (circles).  $T = 20^\circ\text{C}$ . The solid lines represent the fitted straight lines to the data points.  $I(Q) \sim \exp(-Q^2 R_g^2/3)$  where  $R = R_g * \sqrt{5/3}$  is the droplet radius.

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