Contents lists available at ScienceDirect

Bioorganic & Medicinal Chemistry Letters





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Sulfonamide inhibition profile of the γ -carbonic anhydrase identified in the genome of the pathogenic bacterium *Burkholderia pseudomallei* the etiological agent responsible of melioidosis



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ARTICLE INFO

Article history: Received 27 October 2016 Revised 7 December 2016 Accepted 10 December 2016 Available online 12 December 2016

Keywords: Carbonic anhydrase γ-Class Inhibitor Sulfonamide Sulfamate Acetazolamide Drug resistance Burkholderia pseudomallei

ABSTRACT

A new γ -carbonic anhydrase (CA, EC 4.1.1.1) was cloned and characterized kinetically in the genome of the bacterial pathogen *Burkholderia pseudomallei*, the etiological agent of melioidosis, an endemic disease of tropical and sub-tropical regions of the world. The catalytic activity of this new enzyme, BpsCA γ , is significant with a k_{cat} of $5.3 \times 10^5 \text{ s}^{-1}$ and k_{cat}/K_m of $2.5 \times 10^7 \text{ M}^{-1} \times \text{s}^{-1}$ for the physiologic CO₂ hydration reaction. The inhibition constant value for this enzyme for 39 sulfonamide inhibitors was obtained. Acetazolamide, benzolamide and metanilamide were the most effective (K₁s of 149–653 nM) inhibitors of BpsCA γ activity, whereas other sulfonamides/sulfamates such as ethoxzolamide, topiramate, sulpiride, indisulam, sulthiame and saccharin were active in the micromolar range (K₁s of 1.27–9.56 µM). As *Burkholderia pseudomallei* is resistant to many classical antibiotics, identifying compounds that interfere with crucial enzymes in the *B. pseudomallei* life cycle may lead to antibiotics with novel mechanisms of action.

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In all living organisms, the interconversion of CO_2 and HCO_3^- is balanced naturally to maintain the equilibrium between dissolved inorganic carbon dioxide (CO_2), carbonic acid (H_2CO_3), bicarbonate (HCO_3^-) and carbonate (CO_3^{2-}). All these molecules have a pivotal function in all lifeforms. For example, CO_2 is fixed as 3-phosphoglycerate (3-PGA) in plants by the photosynthetic C3 cycle, through the action of RuBisCo,^{1,2} which is the only enzyme capable of net carbon assimilation in the chloroplast stroma; HCO_3^- is the substrate of phosphoenolpyruvate carboxylase (PEPC), which is the enzyme involved in the formation of oxaloacetate in the mesophyll cells of the C4 photosynthetic plants³; moreover, HCO_3^- is the main molecule used by other carboxylating enzymes involved in biosynthetic pathways of fatty acids, amino acids and nucleotides in animals. Thus, the HCO_3^-/CO_3^{2-} ratio is fundamental to buffering the pH in many tissues and various organisms. For all organisms and tissues, the spontaneous hydration/dehydration reaction (1) is very slow at physiological pH.⁴

$$\mathrm{CO}_2 + \mathrm{H}_2\mathrm{O} \leftrightarrows \mathrm{H}\mathrm{CO}_3^- + \mathrm{H}^+ \tag{1}$$

To face this problem, all life forms evolved a specific family of metalloenzymes, the carbonic anhydrases (CAs, EC 4.2.1.1), with the primary function to strongly accelerate the interconversion of CO₂ and HCO₃.^{5–9} This superfamily includes seven distinct classes known as the α -, β -, γ -, δ -, ζ -, η - and θ -CAs.^{10–18} These enzymes are involved in many physiologic processes, such as photosynthesis, respiration, CO₂ transport, as well as metabolism of xenobiotics (e.g., cyanate in *Escherichia coli*). Some of the catalytically active α - and θ -CAs can also catalyze the hydrolysis of esters, such as 4-nitrophenyl acetate (4-NpA). However, no esterase activity was detected so far for enzymes belonging to the other five classes (β -, γ -, δ -, ζ -, and η -CAs).^{11–13,16,19–21}

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Bacteria encode for enzymes belonging to α -, β - and γ -CA classes. α - and γ -CAs contain zinc ion (Zn²⁺) in their active site, coordinated by three histidine residues and a water molecule/ hydroxide ion, whereas in β -CAs the zinc ion is coordinated by two Cys and one His residues. In the search for antibiotics with a novel mechanism of action, a large number of CAs have been investigated in detail in pathogenic bacteria such as Helicobacter pylori, Brucella suis, Mycobacterium tuberculosis, Streptococcus pneumoniae, Salmonella enterica, Vibrio cholerae, Legionella pneumophila, Porphyromonas gingivalis, Streptococcus mutans and others.^{9,16,17,22} It has been shown that the use of inhibitors of this family of metalloenzymes leads to growth defects of the pathogen. Thus, it is feasible that new strategies for the development of antiinfectives with a new and less explored mechanism of action can be developed. The classical CA inhibitors (CAIs) are the primary sulfonamides. RSO₂NH₂, which have been in clinical use for more than 60 years as diuretics, and systemically acting antiglaucoma drugs.^{23,24} For example, there are several clinically used drugs (or agents in clinical development) belonging to the sulfonamide or sulfamate classes, which show significant CAI inhibitory activity (Fig. 1).^{1,25–32} However, recently sulfonamide/sulfamate CAIs have emerged for targeting the bacterial proteins and have the potential for use as antiinfectives.^{17,21,25,27,33–40} In Fig. 1, a large number of sulfonamides/sulfamates are shown that are often used for the screening of the inhibition profile of various CAs, including bacterial enzymes belonging to all three classes (α , β and γ).^{6,8,9,11,13,14,41–44}

Among the Gram-negative saprophytic bacteria living in water and soil, *Burkholderia pseudomallei* is the etiological agent responsible for melioidosis, which is an endemic disease of tropical and sub-tropical regions.⁴⁵ Humans can be infected by skin inoculation, inhalation, and ingestion of contaminated water and soil. Infected people can manifest symptoms, such as bacteremia with septic shock, pneumonia, pericarditis, genitourinary infections, inflammation of parotid glands, central nervous system infection, septic arthritis, abscess involving spleen, liver and adrenal glands.⁴⁵

We investigated the presence of CAs in the genome of *B. pseudo-mallei* due to the following reasons: *i*) *Burkholderia pseudomallei* is



Fig. 1. Sulfonamides/sulfamates that were investigated as CAIs.



Fig. 1 (continued)

fundamentally resistant to penicillin, ampicillin, first-generation and second-generation cephalosporins, macrolides, quinolones and most aminoglycosides; *ii*) it has been reported that the optimal proliferation temperature of this bacterium is around 40 °C in neutral or slightly acidic conditions (pH 6.8–7.0), which might be controlled by CAs, which are enzymes that are involved in the pH homeostasis in all living organisms.⁴⁶

Owing to the limited therapeutic options for treating *Burkholderia pseudomallei* induced infections and the pivotal role of CAs in pH homeostasis, targeting the CA family is a promising to discover new antibacterials devoid of resistance problems. *B. pseudomallei* genome encodes for at least three β -CAs and one γ -CA, but not for α -class enzymes. Here, we report the inhibition profiles of BpsCA γ in comparison to those obtained for human CAs (α -CAs) and other bacterial γ -CAs using the sulfonamides/sulfamates that are shown in Fig. 1. This study may be of interest for designing new types of inhibitors that may have clinical applications.

In Table 1, the rate constants and related kinetic values (k_{cat} , K_M and k_{cat}/K_M) for the γ -CA identified in the genome of *B. pseudomallei*, and the inhibition constant values (K_I) for the inhibitor acetazolamide (5-acetamido-1,3,4-thiadiazole-2-sulfonamide) and BpsCA γ are shown. These data were compared to the kinetic parameters of other CAs belonging to α -, β - and γ -classes identified in different organisms. The catalytic activity of these enzymes was determined using the 'stopped-flow' technique.⁴⁷ The kinetic parameters for BpsCA γ were: k_{cat} of 5.3 \times 10⁵ s⁻¹ and k_{cat}/K_m of

 $2.5 \times 10^7 \text{ M}^{-1} \times \text{s}^{-1}$ (Table 1). BpsCA γ possesses a moderate CO₂ hydrase activity when compared with the human isoform hCA I, which is similar to the bacterial γ -CAs from various pathogenic bacteria (*Porphyromonas gingivalis*, an oral cavity pathogenic bacterium; *Vibrio cholerae*, etiological agent of cholera; Table 1).^{10,13,30,34,42,48-50} Furthermore, the BpsCA γ activity was effectively inhibited (K₁ of 149 nM) by the clinically used sulfonamide inhibitor acetazolamide.

To determine the hydratase activity of the BpsCA γ , this enzyme was analyzed using protonographic analysis,^{51–53} which is a new technique in which pH changes resulting from CAs on SDS–PAGE can be determined colorimetrically. In Fig. 2, protonograms obtained using the commercially available bovine bCA (α -CA) and the recombinant BpsCA γ are shown.

The protonogram of bCA showed a single band of activity corresponding to a monomer of 30 kDa corresponding to the mass of the monomer bCA and consistent with mammalian α -CAs being active as monomers (Fig. 2).^{21,22,44,54–56} The protonographic analysis of BpsCA γ showed a band of activity at the molecular weight of 22 kDa. It has been reported that γ -CAs catalyze the hydration of carbon dioxide to bicarbonate and protons when the γ -CAs monomers assemble into a trimer to form a triangular motif with a real molecular weight of about 65 kDa.^{57,58} The apparent molecular weight of 22 kDa on the SDS-PAGE/protonogram is due to the SDS concentration, which results from the separation of the trimeric state of the protein allowing the migration of the enzyme

Table 1

Comparison of the kinetic parameters for the CO₂ hydration reaction catalyzed by BpsCA γ , the human cytosolic isozymes hCA I and II (α -class CAs) and the γ -CAs from *Vibrio cholerae* (VchCA γ) and *P. gingivalis* (PgiCA γ). Acetazolamide inhibition data are also shown. Human isoenzymes were analyzed at 20 °C and pH 7.5 in 10 mM HEPES buffer and 20 mM Na₂SO₄, while the bacterial enzymes were measured at 20 °C, pH 8.3 in 20 mM TRIS buffer and 20 mM Na₂CO₄.

Enzyme	Activity level	Class	$k_{cat} \left(s^{-1} \right)$	$k_{cat}/K_m (M^{-1} s^{-1})$	K _I (acetazolamide) (nM)
hCA I	Moderate	α	2.0×10^5	$5.0 imes10^7$	250
hCA II	Very high	α	$1.4 imes10^6$	$1.5 imes 10^8$	12
VchCAγ	Moderate	γ	$7.39 imes 10^5$	$6.4 imes 10^7$	473
PgiCAγ	Moderate	γ	$4.1 imes 10^5$	$5.4 imes 10^7$	324
BpsCAγ	Moderate	γ	$\textbf{5.3}\times \textbf{10}^{5}$	$2.5 imes 10^7$	149



Fig. 2. Protonographic analysis of BpsCA_Y. The gel was loaded with bCA and BpsCA_Y. STD corresponds to standards and the yellow band denotes CA activity due to the change of color of the pH indicator.

as a monomer (see Fig. 2). The yellow band found in correspondence of the inactive monomeric form of BpsCA γ is expected because at the end of the electrophoretic run, the SDS is removed from the gel. This procedure led to the rearrangement of γ -CA monomers in the gel and the final result is the reconstitution of the active trimeric form of the γ -CA.^{52,53}

Here, we report the sulfonamide/sulfamate inhibition study of BpsCA γ comparing it with data obtained for the human cytosolic isozymes hCA I and II (α -class CAs) and the γ -CAs from *Vibrio cholerae* (VchCA γ). The following structure-activity relationship (SAR) may be concluded by considering the inhibition data in Table 2:

- (i) BpsCA γ is less sensitive to sulfonamide CAIs compared to other α (e.g., hCA II) or γ -CAs (e.g., VchCA γ) investigated earlier,⁸ for which many nanomolar (and low nanomolar) inhibitors were found (Table 2). It may be observed that only three sulfonamides had K_Is in the range of 149–653 nM, all the other ones having inhibition constants >1 μ M. The most effective BpsCA γ inhibitors were acetazolamide **AAZ**, benzolamide **BZA**, and metanilamide **1**. They incorporate both aromatic (1) and heterocyclic (**AAZ**, **BZA**) sulfonamide motifs. However, as discussed below, small changes in their scaffolds can result in a dramatic decrease in inhibitory activity.
- (ii) Moderate to low BpsCAγ inhibitory activity was observed for the following derivatives: 2, 3, 14, 15, 17–23, MZA, EZA, TPM, SLP, IND, SLT and SAC, which had K₁s in the range of 1270–9560 nM (Table 2). Apart from the clinically used derivatives, which incorporate heterocyclic rings (MZA, EZA), aromatic (SLP, IND, SLT) and sugar (TPM) scaffolds, the secondary, acyl-sulfonamide SAC was also among this

Table 2

Sulfonamides/sulfamates inhibition constants (K_Is, nM) for the human α -CAs (isoforms hCA I and II) and the γ -CAs identified in the genome of V. cholera and B. pseudomallei.

K _I s, nM*							
Inhibitor	hCA I	hCA II	VchCAγ	BpsCAγ			
1	45,400	295	672	574			
2	25,000	240	95.3	1720			
3	28,000	300	93.6	1550			
4	78,500	320	76.3	>50,000			
5	25,000	170	80.6	>50,000			
6	21,000	160	69.0	>50,000			
7	8300	60	73.6	>50,000			
8	9800	110	73.6	12,500			
9	6500	40	95.3	>50,000			
10	6000	70	544	>50,000			
11	5800	63	87.1	14,000			
12	8400	75	563	23,500			
13	8600	60	66.2	18,400			
14	9300	19	69.9	1810			
15	6	2	88.5	9650			
16	164	46	556	10,800			
17	185	50	6223	1825			
18	109	33	5100	1500			
19	95	30	4153	1838			
20	690	12	5570	1810			
21	55	80	764	1335			
22	21,000	125	902	1805			
23	23,000	133	273	1700			
24	24,000	125	73.3	24,500			
AAZ	250	12	473	149			
MZA	50	14	494	1595			
EZA	25	8	85.1	1865			
DCP	1200	38	1230	>50,000			
DZA	50,000	9	87.3	2260			
BRZ	45,000	3	93.0	1270			
BZA	15	9	//.6	653			
IPM	250	10	68.8	3010			
ZNS	50	35	725	>50,000			
SLP	1200	40	//.9	5600			
	31	15	91.3	1800			
	>50,000	43 21	01/	>50,000			
	274	21	034	>50,000			
SAC	5/4 18540	9 5050	404	8900 1550			
JAC	10,040	200	500	1550			
псі	528	290	500	>50,000			

 $^{\ast}\,$ Mean from 3 different assays. Errors in the range of $\pm 10\%$ of the reported values (data not shown).

type of CAIs. Other derivatives in this category are either simple aromatic compounds (sulfanilamide **2**, benzene-1,3disulfonamide **3**, 4-hydroxybenzenesulfonamide **15**, its hydroxyethyl congener **17**, 4-carboxybenzenesulfonamide **18**) or incorporate more elaborated scaffolds (as in **19–23**) that are prevalent for the sulfonylated sulfonamide type (e.g., aminobenzolamide **20** and benzolamide **BZA** are typical examples of such a scaffold). However, relatively minor changes in the scaffold results in a dramatic change in the CA inhibitory activity. For example, the isomers **1** and **2** that differ in the position of the amino group with respect to the

sulfonamide, differ in a factor of 3 in their BpsCA γ inhibitory activity. In the structurally related series 15-17, the compound incorporating the longer spacer (n = 2, 17) inhibited BpsCAy more effectively by over a factor of 5 than the compound **15** that had a shorter spacer (n = 0). However, the derivative with the hydroxymethyl moiety 16 was a much weaker inhibitor, with a K_I of 10.8 μ M. Other such examples are abundant in this series of compounds that was investigated, with the deprotected acetazolamide 13 being less effective as an inhibitor than AAZ by over a factor of 100. In contrast, for methazolamide and the deacetylated methazolamide (14) pair, the difference of activity is minor, with **14** being only 1.13 times weaker as an inhibitor compared to MZA (Table 2). All these data are indicative of the fact that the recognition between the enzyme active site and the inhibitor molecules is governed by multiple factors that are challenging to elucidate without an X-ray crystal structure of the enzyme or high-resolution hydrogen deuterium exchange data. In fact, up until now, a very limited number of γ -CAs have been crystallized, and no crystal structures have been reported for a γ -CA and inhibitor complex.^{57–60}

- (iii) The inhibitory action against BpsCA γ was very low for the following compounds: 8, 11-13, 16 and 24, which had K_Is in the range of 10,800-24,500 nM. Again, they belong to heterogeneous classes of sulfonamides, with aromatic monosulfonamides (8, 16, 24), disulfonamides (11, 12), and a heterocylic sulfonamide derivative, deacetylated acetazolamide **13** (Table 1).
- (iv) A number of the investigated sulfonamides did not inhibit BpsCA γ up to 50 μ M, which is the maximum concentration of inhibitor in the assay system. They are: 4–7, 9, 10, DCP, ZNS, VLX, CLX and HCT. Whereas some of these compounds have rather bulky, complicated scaffold (VLX, CLX and HCT), which could be sterically hindered from binding within the rather shallow γ -CA active site, the simple derivatives **4**–**7**, **9** and **10** should not experience such issues with binding. It is thus obvious that SAR for the inhibition of this enzyme is rather complex and poorly understood with the available data at this moment.
- (v) The inhibition profile of BpsCA γ is very different from that of other α - or γ -CAs of mammalian or bacterial origin, making this enzyme a peculiar case, which deserves further investigation.

In conclusion, we have cloned and characterized the kinetic profile of a new γ -CA in the genome of the bacterial pathogen Burkholderia pseudomallei, the etiological agent of melioidosis, an endemic disease of tropical and sub-tropical regions of the world. For the physiologic CO₂ hydration reaction, this new enzyme, BpsCA_γ, exhibits significant catalytic activity (k_{cat} of 5.3 \times 10⁵ s⁻¹ and k_{cat}/K_m of $2.5 \times 10^7 \text{ M}^{-1} \times \text{s}^{-1}$). Few sulfonamides were effective BpsCA γ inhibitors, with acetazolamide, benzolamide and metanilamide being the most effective inhibitors identified (K₁s of 149–653 nM) whereas other sulfonamides/sulfamates such as ethoxzolamide, topiramate, sulpiride, indisulam, sulthiame and saccharin were less effective (K_Is of 1.27-9.56 µM). As Burkholderia pseudomallei is resistant to many classical antibiotics, discovering compounds that interfere with crucial enzymes in its life cycle may lead to the development of antibiotics with novel mechanisms of action.

Acknowledgment

This work was supported in part by the Distinguished Scientists Fellowship Program (DSFP) from King Saud University, Saudi Arabia.

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and 20 mM NaClO₄ (for maintaining constant the ionic strength), following the initial rates of the CA-catalyzed CO₂ hydration reaction for a period of 10-100 s. The CO₂ concentrations ranged from 1.7 to 17 mM for the determination of the kinetic parameters (by Lineweaver-Burk plots) and inhibition constants. For each inhibitor at least six traces of the initial 5-10% of the reaction have been used for determining the initial velocity. The uncatalyzed rates were determined in the same manner and subtracted from the total observed rates. Stock solutions of inhibitor (10-100 mM) were prepared in distilleddeionized water and dilutions up to 0.01 mM were done thereafter with the assay buffer. Inhibitor and enzyme solutions were preincubated together for 15 min at room temperature prior to assay, in order to allow for the formation of the E-I complex or for the eventual active site mediated hydrolysis of the inhibitor. The inhibition constants were obtained by non-linear least-squares methods using PRISM 3 and the Cheng-Prusoff equation, as reported earlier, and represent the mean from at least three different determinations. All CA isoforms were recombinant ones obtained in-house. All salts/small molecules were of the highest purity available, from Sigma-Aldrich (Milan, Italy).

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